

Tris(ethylenediamonium) bis[(2-aminoethyl)ammonium] bis[bis(μ_5 -hydrogen phosphato)penta- μ_2 -oxido-decaoxido-pentamolybdenum(VI)] decahydrate

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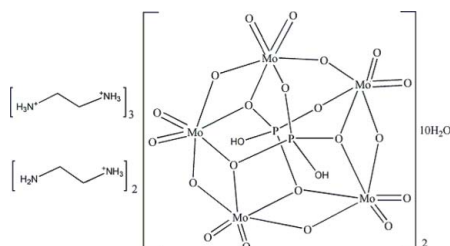
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.012$ Å; R factor = 0.041; wR factor = 0.114; data-to-parameter ratio = 13.3.

The title compound, $(\text{C}_2\text{H}_{10}\text{N}_2)_3(\text{C}_2\text{H}_9\text{N}_2)_2[\text{Mo}_5(\text{HPO}_4)_2\text{O}_{15}]\cdot 10\text{H}_2\text{O}$, was prepared under hydrothermal conditions at pH 5.0. The structure contains mono- and diprotonated ethylenediamine cations, $[\text{Mo}_5\text{O}_{15}(\text{HPO}_4)_2]^{4-}$ anions and uncoordinated water molecules. The $[\text{Mo}_5\text{O}_{15}(\text{HPO}_4)_2]^{4-}$ heteropolyoxometallate anion is made up of five MoO_6 octahedra sharing an edge and forming a ring, which is closed by common corners of the terminal MoO_6 octahedron. The ring is topped on both sides by two slightly distorted PO_4 tetrahedra, sharing three corners with three MoO_6 octahedra. The terminal oxygen atoms of the PO_4 units are protonated. Together with the anions, the water molecules and the ethylenediamonium cations are involved in $\text{N}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonding, forming a three-dimensional supramolecular network.

Related literature

For background to polyoxometalates, see: Coronado & Gomez-Garcia (1998); Niu *et al.* (2009); Ruether *et al.* (2003). For the structure of $(\text{C}_2\text{H}_{10}\text{N}_2)_2[\text{Mo}_5\text{O}_{15}(\text{HPO}_4)_2]$, see: Sun *et al.* (2003). For structures containing the $[\text{Mo}_5\text{O}_{15}(\text{PO}_4)_2]^{6-}$ anion, see: Gong *et al.* (2006); Skibsted *et al.* (2000). For the bond-valence method, see: Brown (2002).



Experimental

Crystal data

$(\text{C}_2\text{H}_{10}\text{N}_2)_3(\text{C}_2\text{H}_9\text{N}_2)_2\cdot$
 $[\text{Mo}_5(\text{HPO}_4)_2\text{O}_{15}]\cdot 10\text{H}_2\text{O}$
 $M_r = 2312.06$
 Triclinic, $P\bar{1}$
 $a = 10.0045$ (11) Å
 $b = 10.6625$ (12) Å
 $c = 15.1884$ (19) Å
 $\alpha = 87.405$ (2)°

$\beta = 73.119$ (1)°
 $\gamma = 77.978$ (1)°
 $V = 1516.2$ (3) Å³
 $Z = 1$
 Mo $K\alpha$ radiation
 $\mu = 2.23$ mm⁻¹
 $T = 298$ K
 $0.38 \times 0.34 \times 0.30$ mm

Data collection

Siemens SMART CCD area-detector diffractometer
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.485$, $T_{\max} = 0.554$

7582 measured reflections
 5253 independent reflections
 4015 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.114$
 $S = 1.04$
 5253 reflections
 396 parameters

5 restraints
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 1.28$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.07$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N5}-\text{H5E}\cdots\text{O20}^{\text{i}}$	0.90	2.66	3.075 (8)	109
$\text{N5}-\text{H5E}\cdots\text{O28}^{\text{ii}}$	0.90	2.01	2.796 (9)	144
$\text{N5}-\text{H5D}\cdots\text{O10}$	0.89	2.45	3.069 (8)	127
$\text{N5}-\text{H5D}\cdots\text{O6}$	0.89	2.01	2.846 (8)	156
$\text{N5}-\text{H5C}\cdots\text{O21}^{\text{i}}$	0.89	2.17	3.046 (8)	170
$\text{N4}-\text{H4E}\cdots\text{O1}^{\text{i}}$	0.89	1.93	2.806 (8)	167
$\text{N4}-\text{H4D}\cdots\text{O12}^{\text{iii}}$	0.89	2.65	3.357 (8)	137
$\text{N4}-\text{H4D}\cdots\text{O22}^{\text{iii}}$	0.89	2.60	3.099 (8)	116
$\text{N4}-\text{H4D}\cdots\text{O4}^{\text{iii}}$	0.89	2.08	2.907 (8)	155
$\text{N4}-\text{H4C}\cdots\text{O25}$	0.89	1.92	2.803 (8)	171
$\text{N3}-\text{H3D}\cdots\text{O17}^{\text{i}}$	0.87	2.25	3.117 (8)	176
$\text{N3}-\text{H3C}\cdots\text{O15}^{\text{iv}}$	0.89	1.87	2.732 (7)	162
$\text{N2}-\text{H2E}\cdots\text{O23}^{\text{iii}}$	0.90	2.56	3.030 (8)	113
$\text{N2}-\text{H2E}\cdots\text{O5}^{\text{v}}$	0.90	1.84	2.699 (8)	159
$\text{N2}-\text{H2D}\cdots\text{O20}^{\text{v}}$	0.89	2.14	2.924 (8)	146
$\text{N2}-\text{H2C}\cdots\text{O6}^{\text{iii}}$	0.90	2.49	3.310 (8)	151
$\text{N2}-\text{H2C}\cdots\text{O12}^{\text{iii}}$	0.90	2.35	3.084 (8)	139
$\text{N1}-\text{H1C}\cdots\text{O7}^{\text{iv}}$	0.90	2.46	3.259 (8)	149
$\text{N1}-\text{H1C}\cdots\text{O16}^{\text{iv}}$	0.90	2.28	3.011 (8)	138
$\text{N1}-\text{H1B}\cdots\text{O28}^{\text{vi}}$	0.89	1.93	2.819 (8)	173
$\text{N1}-\text{H1A}\cdots\text{O5}^{\text{v}}$	0.90	1.92	2.772 (8)	159
$\text{O28}-\text{H28B}\cdots\text{O23}^{\text{vii}}$	0.86	2.39	3.157 (8)	149
$\text{O28}-\text{H28A}\cdots\text{O27}$	0.84	2.31	2.740 (10)	112
$\text{O27}-\text{H27B}\cdots\text{O17}^{\text{i}}$	0.87	2.46	2.912 (10)	113
$\text{O27}-\text{H27B}\cdots\text{O22}^{\text{vii}}$	0.87	2.11	2.916 (10)	155
$\text{O27}-\text{H27A}\cdots\text{O10}^{\text{viii}}$	0.87	2.03	2.875 (9)	163
$\text{O26}-\text{H26B}\cdots\text{O19}^{\text{j}}$	0.84	2.40	3.064 (8)	136
$\text{O26}-\text{H26B}\cdots\text{O17}^{\text{i}}$	0.84	2.36	2.874 (8)	120
$\text{O26}-\text{H26A}\cdots\text{O14}$	0.84	2.11	2.858 (8)	148
$\text{O25}-\text{H25B}\cdots\text{O21}^{\text{i}}$	0.84	1.97	2.808 (7)	170
$\text{O25}-\text{H25B}\cdots\text{O4}^{\text{i}}$	0.84	2.57	3.083 (7)	120
$\text{O25}-\text{H25A}\cdots\text{O11}$	0.85	1.93	2.745 (7)	163
$\text{O24}-\text{H24B}\cdots\text{O25}^{\text{viii}}$	0.86	2.08	2.868 (9)	151
$\text{O24}-\text{H24A}\cdots\text{O1}^{\text{iv}}$	0.86	1.97	2.795 (8)	159
$\text{O5}-\text{H5F}\cdots\text{O28}^{\text{ii}}$	0.84	2.02	2.845 (8)	168
$\text{O1}-\text{H1F}\cdots\text{N3}^{\text{ix}}$	0.85	2.18	2.766 (8)	126

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x + 1, -y + 1, -z$; (iii) $-x + 1, -y + 2, -z + 1$; (iv) $-x + 1, -y + 1, -z + 1$; (v) $x - 1, y, z + 1$; (vi) $x, y, z + 1$; (vii) $x - 1, y - 1, z$; (viii) $x, y - 1, z$; (ix) $x + 1, y, z$.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINTE* (Siemens, 1996); data reduction: *SAINTE*; program(s) used to solve

structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2326).

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